

Ultraviolet-Light-Induced Reversible and Stable Carrier Modulation in MoS₂ Field-Effect Transistors

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The tuning of charge carrier concentrations in semiconductor is necessary in order to approach high performance of the electronic and optoelectronic devices. It is demonstrated that the charge-carrier density of single-layer (SL), bilayer (BL), and few-layer (FL) MoS₂ nanosheets can be finely and reversibly tuned with N₂ and O₂ gas in the presence of deep-ultraviolet (DUV) light. After exposure to N₂ gas in the presence of DUV light, the threshold voltages of SL, BL, and FL MoS₂ field-effect transistors (FETs) shift towards negative gate voltages. The exposure to N₂ gas in the presence of DUV light notably improves the drain-to-source current, carrier density, and charge-carrier mobility for SL, BL, and FL MoS₂ FETs. Subsequently, the same devices are exposed to O₂ gas in the presence of DUV light for different periods and the electrical characteristics are completely recovered after a certain time. The doping by using the combination of N₂ and O₂ gas with DUV light provides a stable, effective, and facile approach for improving the performance of MoS₂ electronic devices.

Recently, many other 2D materials, such as the transition-metal dichalcogenides, transition-metal oxides and boron nitride, have been investigated.^[10–12] Molybdenum disulfide (MoS₂), a layered transition-metal dichalcogenide n-type semiconductor with an indirect bandgap of 1.2 eV (where single-layer (SL) MoS₂ has a direct bandgap of 1.8 eV and bilayer (BL) MoS₂ has an indirect bandgap of 1.3 eV), is attracting increasing interest for its novel electronic and optoelectronic properties.^[13–16] Bulk MoS₂ crystal has a structure of interacting layers of a S-Mo-S covalently bonded hexagonal quasi-2D network held together by van der Waals interactions.^[17] Thus, SL, BL or few-layer (FL) MoS₂ can be easily fabricated in a similar manner to graphene by mechanical exfoliation.

1. Introduction

The identification of graphene as the first two-dimensional (2D) material with exotic electronic, optical and mechanical properties has given birth to research on 2D nanomaterials.^[1,2] Graphene has been intensively investigated over the last decade because of its interesting physical properties, chemical properties and technological applications in nanoelectronics, optoelectronics, energy harvesting and sensors.^[3–9] Graphene has a conical Dirac spectrum and linear dispersion characteristic.^[2] However, the absence of an energy gap in the electronic band structure limits its practical applications in nanoelectronics and optoelectronics.

Nowadays, MoS₂ is incredibly attractive for use in next-generation nanoelectronics and optoelectronics devices because it is a very thin 2D material having a semi-conducting nature, an absence of dangling bonds, thermal stability up to 1100 °C, stability in air, no short-channel effect, mobility comparable to that of silicon (in the range of 100–600 cm² V^{−1} s^{−1}) and a very high on/off current ratio (~10⁸).^[14,18] MoS₂-based devices such as field-effect transistors (FETs), integrated circuits, photodetectors, memory devices, chemical sensors and supercapacitors have already been demonstrated.^[19–26] Among these applications, the FET is the most important and is a basic element of any electronic circuit/device. Presently, researchers are trying to improve the performance of MoS₂ FETs by using different metal contacts (source and drain electrodes), scaling down the device dimensions, using high-K materials for the gate dielectric and doping of MoS₂ nanoflakes with self-assembled monolayers or molecules, or with metal nanoparticles.^[27–34] It is also reported that doping of two-dimensional materials also enhanced the optical properties and very useful for biological applications.^[35,36,41] Doping is the most effective and easy technique for tuning the electrical and optical properties of semiconducting materials. The controlled doping in semiconductor is not only essential for modulating the carrier concentrations and electronic properties but also demanding for fabrication of various complex devices.^[42] The reversible and controlled doping of MoS₂ layers, which is of fundamental importance for FET devices, has not yet been well explored.

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Here, we present charge-carrier modulation in SL, BL, and FL MoS₂ by N₂ and O₂ gas in the presence of deep-ultraviolet (DUV) light. First, SL, BL and FL MoS₂ are exposed to N₂ gas in the presence of DUV light for different periods. The DUV-light-induced changes in SL, BL, and FL MoS₂ in the presence of N₂ and O₂ gas are investigated by making electrical-transport measurements and performing Raman spectroscopy. The shifting of the threshold voltage towards the negative gate voltage after exposure to N₂ gas in the presence of DUV light reveals the n-doping of SL, BL, and FL MoS₂ FETs. After 30 min exposure to N₂ gas in the presence of DUV light, we expose the same devices to O₂ gas in the presence of DUV light for different periods and observe that the devices completely recover after a certain time. We have found that the recovery time is different for SL, BL, and FL MoS₂ FETs. DUV light in the presence of N₂ gas notably improves the drain-to-source current, carrier density and charge-carrier mobility of SL, BL, and FL MoS₂ FETs.

2. Results and Discussion

We start the fabrication process with the transfer of mechanically exfoliated MoS₂ layers (SL, BL and FL MoS₂) using Scotch tape on a 300-nm SiO₂ substrate with underlying highly p-doped silicon. SL, BL, and FL MoS₂ layers were identified with an optical microscope and through Raman spectroscopy and atomic force microscopy (AFM). **Figure 1** shows optical images of SL, BL, and FL MoS₂ layers and their Raman spectra. The different contrast of MoS₂ flakes on Si/SiO₂ substrate [with the brown background being the Si/SiO₂ substrate] reveals the different thicknesses. On the basis of contrast, we can label the SL, BL, and FL MoS₂ flakes as shown in Figure 1a–c, respectively. Figure 1d displays typical Raman spectra taken for the exfoliated SL, BL and FL MoS₂ layers. Raman spectra of the SL, BL, and FL MoS₂ layers show strong signals from both in-plane

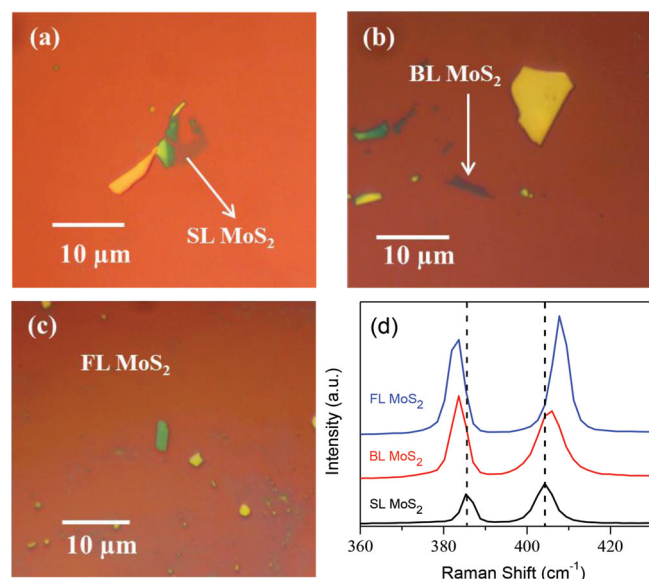


Figure 1. Optical image of the mechanically exfoliated a) SL, b) BL, and c) FL MoS₂ films on Si/SiO₂ substrate. d) Raman spectra of SL, BL, and FL MoS₂ obtained with a 514-nm laser source at room temperature.

E_{1_{2g}} and out-of-plane A_{1g} vibration. The frequency difference between these two Raman modes varies with layer thickness and can be easily used as a “thickness indicator.” The positions of E_{1_{2g}} and A_{1g} are 385.5 and 404.3 cm⁻¹, respectively, for SL MoS₂. The energy difference between the Raman A_{1g} and E_{1_{2g}} modes ($\Delta = A_{1g} - E_{1_{2g}}$) is about 18.8 cm⁻¹, indicating a monolayer. E_{1_{2g}} and A_{1g} are located at 383.6 and 405.2 cm⁻¹, respectively, and the energy difference (Δ) is 21.6 cm⁻¹ for BL MoS₂, confirming the two layers. For FL MoS₂, E_{1_{2g}} and A_{1g} are located at 382.9 and 407.8 cm⁻¹, respectively, and Δ is 24.9 cm⁻¹. Raman spectra for our SL, BL and FL MoS₂ layers are consistent with those reported elsewhere.^[37,38] The thickness of MoS₂ layers was also confirmed by AFM measurements. The thickness of the FL MoS₂ is about 4.6 nm, which indicates seven layers of MoS₂ (Figure S1, Supporting Information).

Figure 2a shows the schematic cross-sectional view of the structure of a MoS₂ FET. The large patterned electrodes (Cr/Au thickness of 5/30 nm) for SL, BL and FL MoS₂ were made employing photolithography on Si/SiO₂ substrate. The source and drain electrodes of transistors were made employing e-beam lithography, evaporation of Cr/Au (6/80 nm) and lift-off processes. Optical images of the devices are presented in Figure S2 (Supporting Information). All devices were annealed at 200 °C for four hours in a flow of 100-sccm Ar and 10-sccm H₂ to remove the residue and contamination. First, we performed electrical characterizations of our devices at room temperature in a vacuum chamber. Figure 2b–d shows the drain current I_{DS} as a function of the applied back-gate voltage V_g at a fixed drain–source voltage, $V_{DS} = 10$ mV, for SL, BL, and FL MoS₂ FETs, respectively. The I_{DS} – V_g graph reveals an n-type channel for all SL, BL, and FL MoS₂ FETs. Repeated V_g sweeps on the same device do not show significant variation, while keeping all the voltages constant results in constant I_{DS} . The relations of drain current versus drain–source voltage (I_{DS} – V_{DS}) at various gate voltages (ranging from –8 to +8 V in steps of 2 V) for SL, BL and FL MoS₂ are given in the insets of the respective figures. The linear drain current versus drain–source voltage (I_{DS} – V_{DS}) characteristic of our devices shows good ohmic contact with source and drain electrodes. I_{DS} of our SL MoS₂ device is lower than those of BL and FL MoS₂ FETs, which is consistent with previously reported results.^[14,18,27] After characterizations of our pristine devices, we exposed the same devices to N₂ gas in the presence of DUV light for different periods. After 30 min exposure to N₂ gas in the presence of DUV light, we exposed the same devices to O₂ gas in the presence of DUV light for different periods. The doping procedure is detailed in the Experimental section. We made electrical-transport and Raman measurements after each exposure. Before each measurement of electrical transport, we vacuumed the sample chamber for half an hour to remove the atmospheric oxygen. The electrical measurements were made in a vacuum by continuously running a vacuum pump. **Figure 3a,c,e** shows I_{DS} as a function of V_g before and after N₂ and O₂ exposure under constant DUV light treatment for different periods. Figure 3a, c and e clearly shows the shifting of the threshold voltage towards a negative gate voltage with N₂ gas in the presence of DUV light for different periods, which revealed n-doping in SL, BL, and FL MoS₂. We tested several devices and observed the same behavior.

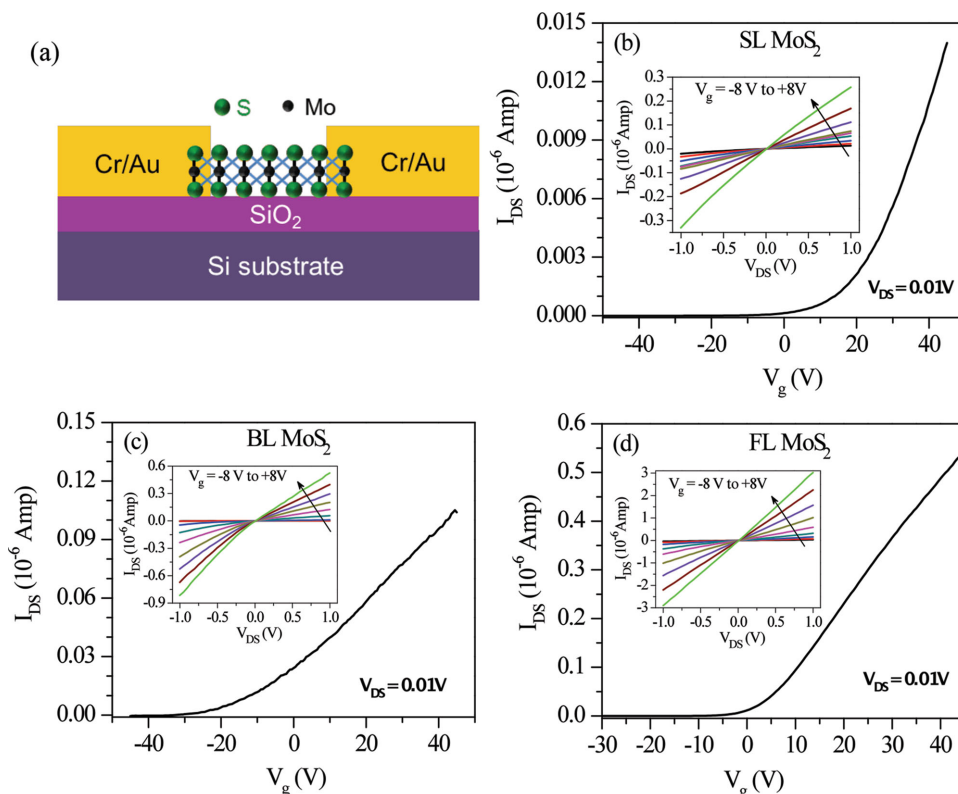


Figure 2. a) Schematic illustration of a MoS₂ transistor. Pristine characteristics of MoS₂ transistors: b) plot of drain current versus back-gate voltage (I_{DS} – V_g) of the SL MoS₂ transistor, with inset showing a plot of drain current versus drain–source voltage (I_{DS} – V_{DS}) at different V_g ranging from –8 to +8 V in steps of 2 V, c) plot of I_{DS} – V_g of the BL MoS₂ transistor, with inset showing a plot of I_{DS} – V_{DS} at different V_g ranging from –8 to +8 V in steps of 2 V, and d) plot of I_{DS} – V_g of the FL MoS₂ transistor, with inset showing a plot of I_{DS} – V_{DS} at different V_g ranging from –8 to +8 V in steps of 2 V.

The n-doping of SL, BL and FL MoS₂ by N₂ gas in the presence of DUV light can be understood in terms of the removal of electron-trapping absorbate groups (such as oxygen or an oxygen-derived group) from MoS₂ nanosheets. The removal of oxygen was also confirmed by energy-dispersive X-ray spectroscopy. The X-rays were scanned on the area of 100 nm × 100 nm on the different locations of all SL, BL, and FL MoS₂ nanosheets. **Table 1** presents the results of the analysis of energy-dispersive X-ray spectroscopy pristine, after DUV+N₂ and after DUV+O₂ treatment for all SL, BL and FL MoS₂ nanosheets. The result of energy-dispersive X-ray spectroscopy shows that significant amount of oxygen was reduced after 30 minute exposure to N₂ gas in the presence of DUV light. However, amount of oxygen was recovered, when we exposed the same devices to O₂ gas in the presence of DUV light for 30 min. The removal of oxygen atoms/molecules from MoS₂ nanosheets significantly increased the drain-to-source current in SL, BL and FL MoS₂ nanosheets. It has been already reported that N₂ gas alone does not affect the electrical properties of MoS₂ nanosheets.^[39] However, in our case, N₂ gas in the presence of DUV light notably affected the electrical properties of MoS₂ nanosheets. The n-doping was also observed by Raman spectroscopy as shown in Figure S3 (Supporting Information). After doping through 30 min exposure to N₂ gas in the presence of DUV light, the A_{1g} peak position shifted towards a lower wave number; the shifting of the A_{1g} peak position of n-doped MoS₂ towards

a lower wave number has already been reported for other systems.^[24,32]

Figure 3b,d,f shows the shifting of the threshold voltage as a function of time of exposure to N₂ and O₂ gas for SL, BL and FL MoS₂, respectively. The threshold voltage of the device is the intercept on the V_g axis obtained by extrapolating the linear portion of the curve of I_{DS} versus V_g to zero I_{DS} . The threshold voltage is found to be –37.4, –40 and –39 V for SL, BL, and FL MoS₂, respectively, after 30 minutes exposure to N₂ gas in the presence of DUV light. We exposed the same devices to O₂ gas in the presence of DUV light for different periods. The threshold voltages shifted towards a positive gate voltage after exposure to O₂ gas in the presence of DUV light, and after a certain period the SL, BL, and FL MoS₂ devices completely recovered. The recovery of MoS₂ devices after exposure to O₂ gas in the presence of DUV may be due to the attachment of oxygen molecules/atoms to MoS₂ nanosheets. There are two possibilities: either oxygen molecules were directly absorbed by MoS₂ nanosheets or oxygen molecules first dissociated into oxygen atoms and the dissociated oxygen atoms then interacted with MoS₂ nanosheets. Oxygen gas molecules can easily dissociate into oxygen atoms under the illumination of DUV light.^[43] The interaction of oxygen molecules/atoms takes an electron from MoS₂ nanosheets so that it is p-doped. The p-doping of MoS₂ nanosheets through exposure to O₂ gas has already been reported.^[40,41]

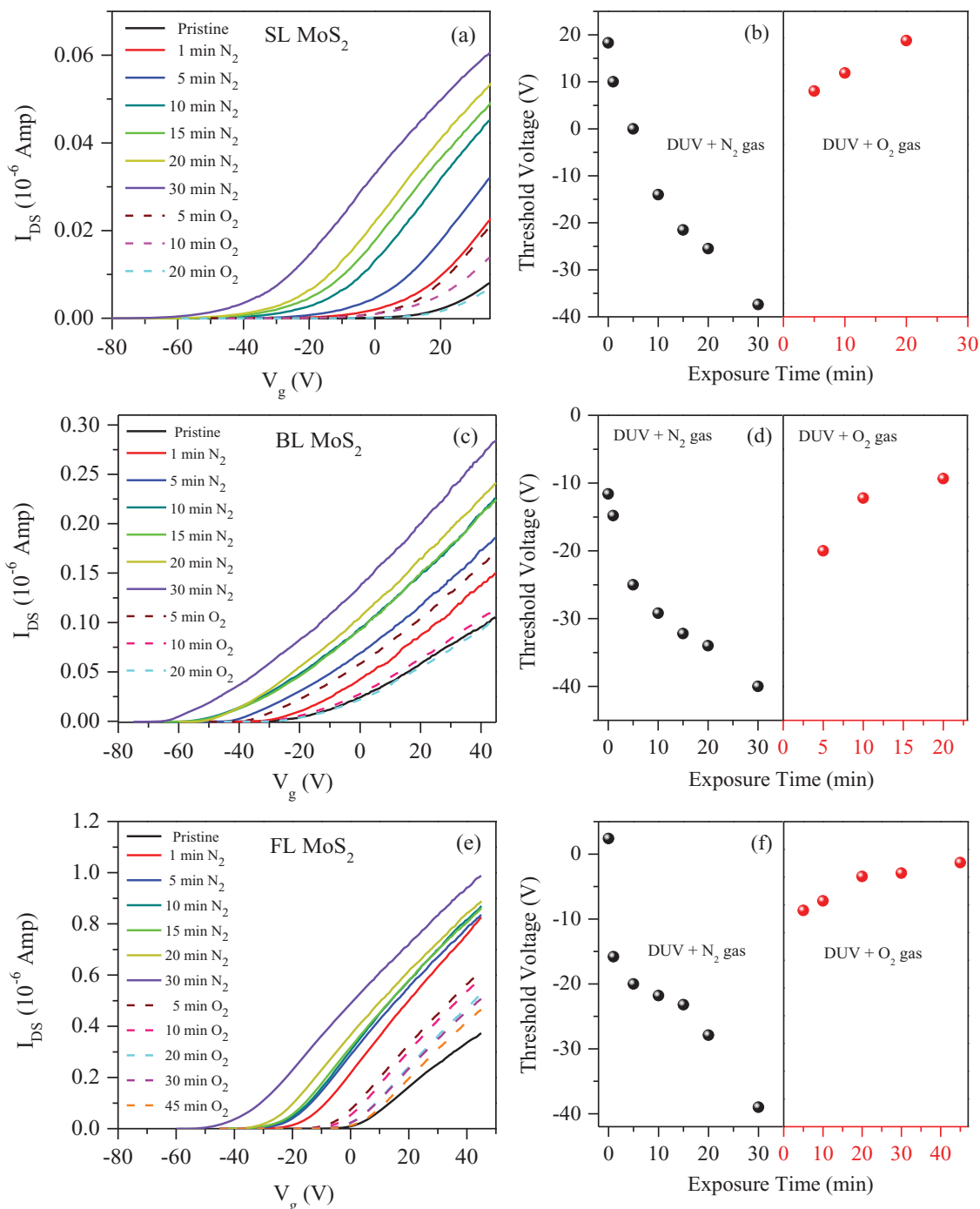


Figure 3. Doping of MoS₂ films by N₂ and O₂ gas in the presence of DUV light. a) Plot of I_{DS} - V_g for different times of exposure to N₂ and O₂ gas in the presence of DUV light and b) the threshold voltage as a function of the time of exposure to N₂ and O₂ gas in the presence of DUV light for SL MoS₂. c) Plot of I_{DS} - V_g for different times of exposure to N₂ and O₂ gas in the presence of DUV light for BL MoS₂. d) the threshold voltage as a function of the time of exposure to N₂ and O₂ gas in the presence of DUV light for BL MoS₂. e) Plot of I_{DS} - V_g for different times of exposure to N₂ and O₂ gas in the presence of DUV light and f) the threshold voltage as a function of the time of exposure to N₂ and O₂ gas in the presence of DUV light for FL MoS₂.

We also measured I_{DS} - V_{DS} characteristics of SL, BL and FL MoS₂ at different gate voltages for each time of exposure to N₂ and O₂ gas in the presence of DUV light. After exposure to N₂ gas in the presence of DUV light, the drain current (I_{DS}) drastically increased with exposure time. After 30 min exposure to N₂ gas in the presence of DUV light, I_{DS} for SL, BL and FL

MoS₂ increased up to ≈ 50 times with respect to pristine drain current at $V_g = +4$ V and $V_{DS} = 0.5$ V. The significant improvement in I_{DS} current is due to the removal of oxygen absorbates from MoS₂ nanosheets and the increase in electron concentrations in the channel region. Another possibility is a higher tunneling probability through the metal-semiconductor interface

Table 1. Energy-dispersive X-ray spectroscopy of SL, BL, and FL MoS₂ nanosheets.

Samples	Element	Pristine	After DUV + N ₂	After DUV + O ₂
SL MoS ₂	O	25.42% ± 0.3%	24.38% ± 0.2%	25.57% ± 0.1%
BL MoS ₂	O	26.26% ± 0.1%	25.33% ± 0.5%	26.38% ± 0.3%
FL MoS ₂	O	23.61% ± 0.4%	23.08% ± 0.06%	23.50% ± 0.5%

barrier due to lowering of the barrier. The reduction in I_{DS} current after exposure to O₂ gas is due to the absorption of oxygen atoms/molecules into sulfur or defect sites on MoS₂ nanosheets and the trapping of electrons. The I_{DS} - V_{DS} characteristics as functions of exposure time to N₂ and O₂ gas in the presence of DUV light at $V_g = +4$ V and $V_{DS} = 0.5$ V for SL, BL, and FL MoS₂ are given in the Supplementary Information (Figure S4, Supporting Information). However, after exposure to N₂ gas in the presence of DUV light, current I_{DS} became weakly dependent on the back-gate voltage (V_g) for SL, BL, and FL MoS₂. This also indicates that SL, BL and FL MoS₂ were heavily n-doped.

Figure 4 shows the mobility and carrier density of SL, BL, and FL MoS₂ as a function of the time of exposure to N₂ and O₂ gas in the presence of DUV light. The mobility of pristine SL, BL and FL MoS₂ is found to be 1.7, 8.1, and 88.8 cm² V⁻¹ s⁻¹, respectively. The mobility of the samples was obtained using the relation $\mu = \frac{L}{C_g W V_{DS}} \left(\frac{\partial I_{DS}}{\partial V_g} \right)$, where L is the channel length, W is the channel width, $\left(\frac{\partial I_{DS}}{\partial V_g} \right)$ is the slope of transfer characteristic of the device, $V_{DS} = 0.01$ V and C_g is the gate capacitance

of ≈ 115 aF μm^{-2} for our Si/SiO₂ substrate. The mobility of our pristine SL, BL, and FL MoS₂ is higher than values previously reported.^[14,21,23,43] The mobilities of SL, BL and FL MoS₂ are significantly improved with exposure to N₂ gas in the presence of DUV light. However, the mobilities of SL, BL, and FL MoS₂ return to the pristine mobility when the devices are exposed to O₂ gas in the presence of DUV light for a certain period. Thus, we can reversibly tune the mobility of MoS₂ through the combination of N₂ and O₂ gas in the presence of DUV light. Figure 4b, d and f shows the carrier density of SL, BL and FL MoS₂ as a function of exposure time to N₂ and O₂ gas in the presence of DUV light. The charge-carrier densities (n) of the devices were estimated using the relation $n = C_g (V_g - V_T)/e$, where V_T is the threshold voltage of the device, $V_g = 30$ V and e is the electronic charge. The carrier densities of SL, BL, and FL MoS₂ notably increase with exposure to N₂ gas in the presence of DUV light and reverse back to the pristine state when the devices are exposed to O₂ gas in the presence of DUV light for a certain period, as shown in Figure 4b,d,f.

The stability of doping of 2D nanomaterials is a major issue nowadays, and important to real device applications. We therefore checked the stability of our SL, BL and FL MoS₂ devices after exposure to N₂ gas in the presence of DUV light. To check the stability of doping of MoS₂ layers induced by N₂ gas in the presence of DUV light, we first exposed SL, BL and FL MoS₂ devices to N₂ gas in the presence of DUV light for 30 min and then measured their electrical characteristics. The electrical characteristics of the same devices were measured again after

15 days in atmospheric environments. **Figure 5a** shows the I_{DS} - V_g plots after 30 min exposure of SL, BL, and FL MoS₂ transistors to N₂ gas in the presence of DUV light and the results for the same devices after 15 days. **Figure 5a** indicates that there were no significant changes in the I_{DS} - V_g plots after 15 days, nor was there much change in the current value of the MoS₂ devices not in threshold voltage. We also checked the stability of the recovered devices. To do so, we first allowed all devices to recover to their pristine state by exposing them to O₂ gas in the presence of DUV light, and then measured their electrical characteristics. The electrical characteristics of the same devices were measured again after 20 days in atmospheric environments. **Figure 5b** shows the I_{DS} - V_g plots for SL, BL, and FL MoS₂ transistors in a pristine state, and the results for the same devices 20 days later. We observed that there was no significant change in their respective current values, which indicates that our devices are very stable.

3. Conclusion

In summary, we presented reversible charge-carrier modulation in different layers of MoS₂ nanosheets through DUV-light-induced doping. The doping of SL, BL, and FL MoS₂ by N₂ and O₂ gas in the presence of DUV light was investigated by measuring charge transport. SL, BL, and FL MoS₂ were exposed to N₂ gas in the presence of DUV light for different periods, and the shifting of the threshold voltage towards a negative gate voltage revealed the n-doping of all SL, BL, and FL MoS₂ FETs. The n-doped MoS₂ nanosheets completely recovered with exposure to O₂ gas in the presence of DUV light. However, recovery times differed for the SL, BL, and FL MoS₂. The DUV-induced doping by N₂ gas molecules significantly improved the performance of MoS₂ FETs and enhanced device parameters such as the drain current, carrier density and charge-carrier mobility for all SL, BL, and FL MoS₂. Reversible carrier modulation by DUV illumination with gas flow is effective and stable. Our study demonstrated a stable and reversible charge-carrier modulation in MoS₂ nanosheets using DUV light. This allows for new possibilities in relation to doping control using DUV light in MoS₂ nanosheets. Our study may also help improve the performance of other electronic and optoelectronic devices based on MoS₂ nanosheets through the application of DUV light exposure.

4. Experimental Methods

Sample Preparation: SL, BL, and FL MoS₂ films were obtained by micromechanical exfoliation of naturally occurring crystals of molybdenite using adhesive tape and then transferring the MoS₂ layers onto a silicon substrate with a 300-nm oxide layer. Before transferring the MoS₂ layers onto the Si/SiO₂ wafer, the Si/SiO₂ wafer was treated with piranha solution (3:1 sulfuric acid to hydrogen peroxide) for 30 min to remove residual organic contaminants. The numbers of layers in the MoS₂ films were identified by optical microscope and Raman spectroscopy, and confirmed by AFM. The FL MoS₂ had seven layers. Raman spectra were collected at room temperature with a Renishaw microspectrometer having a laser wavelength of 514 nm. To avoid local heating and the introduction of defects by the laser, the laser power was kept at ≈ 1.0 mW.

Device Fabrication and Measurements: The large patterned electrodes (Cr/Au thickness of 6/30 nm) for SL, BL and FL MoS₂ films were made

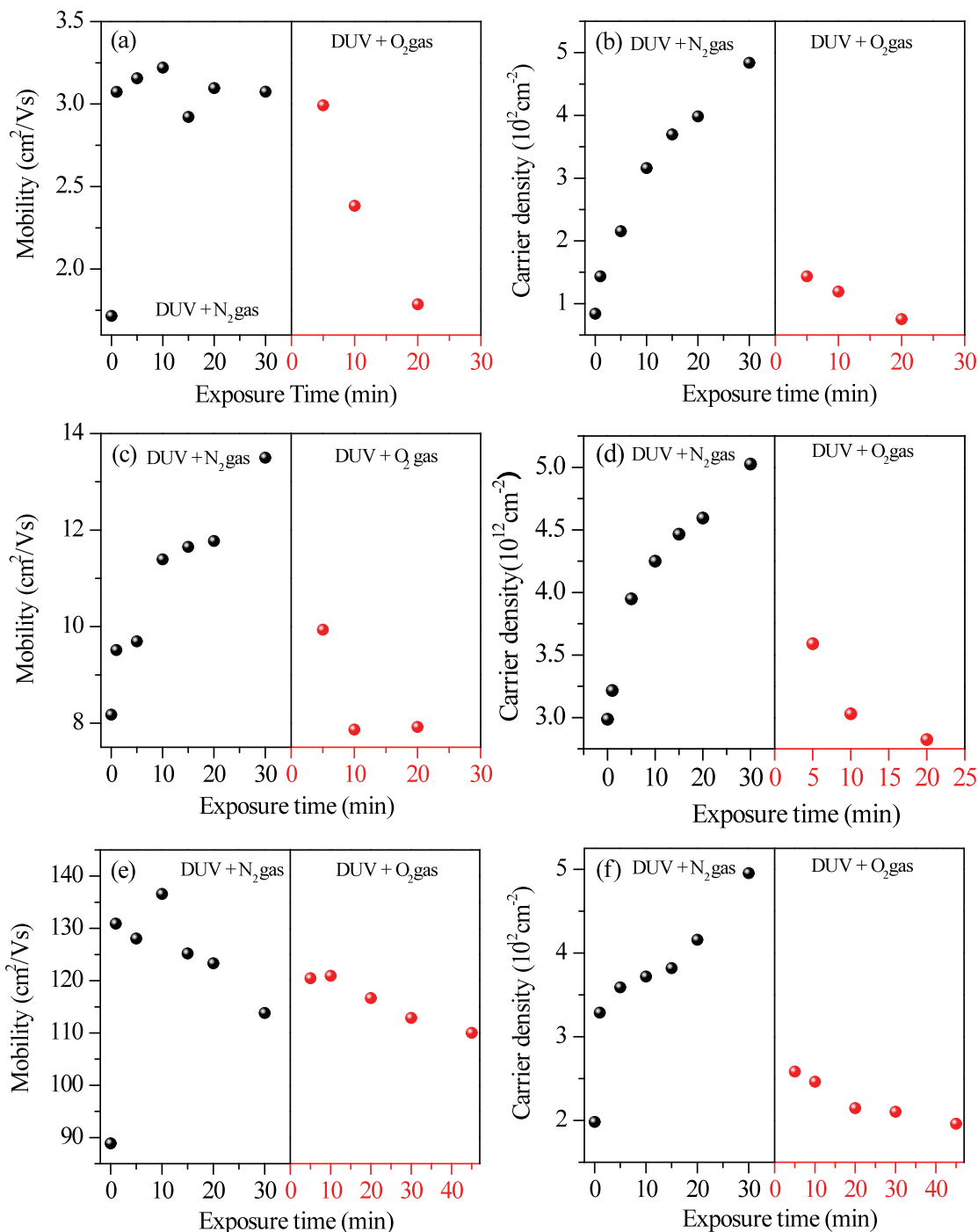


Figure 4. a) Electron mobility and b) charge-carrier density as functions of time of exposure to N₂ and O₂ gas in the presence of DUV light for SL MoS₂. c) Electron mobility and d) charge-carrier density as functions of time of exposure to N₂ and O₂ gas in the presence of DUV light for BL MoS₂. e) Electron mobility and f) charge-carrier density as functions of time of exposure to N₂ and O₂ gas in the presence of DUV light for FL MoS₂.

by photolithography. Source and drain electrodes were patterned employing e-beam lithography and evaporation of Cr/Au (6/80 nm). The channel length of transistors was kept almost constant ($\approx 2.5 \mu\text{m}$) for all devices. After device fabrication, all devices were annealed in a tube furnace at a temperature of 200 °C, in a flow of 100-sccm Ar and 10-sccm H₂ for 4 h to minimize the contact resistance of the devices. The devices were electrically characterized using a Keithley 2400

source meter and Keithley 6485 picometer at room temperature in a vacuum.

Ultraviolet-Light-Induced Doping and Characterizations: SL, BL, and FL MoS₂ films were first exposed to N₂ gas flow in the presence of DUV light ($\lambda = 220 \text{ nm}$, average intensity of 10 mW cm^{-2}) for different periods (maximum of 30 min). The same devices were then exposed to O₂ gas flow for different periods in the presence of DUV light. After

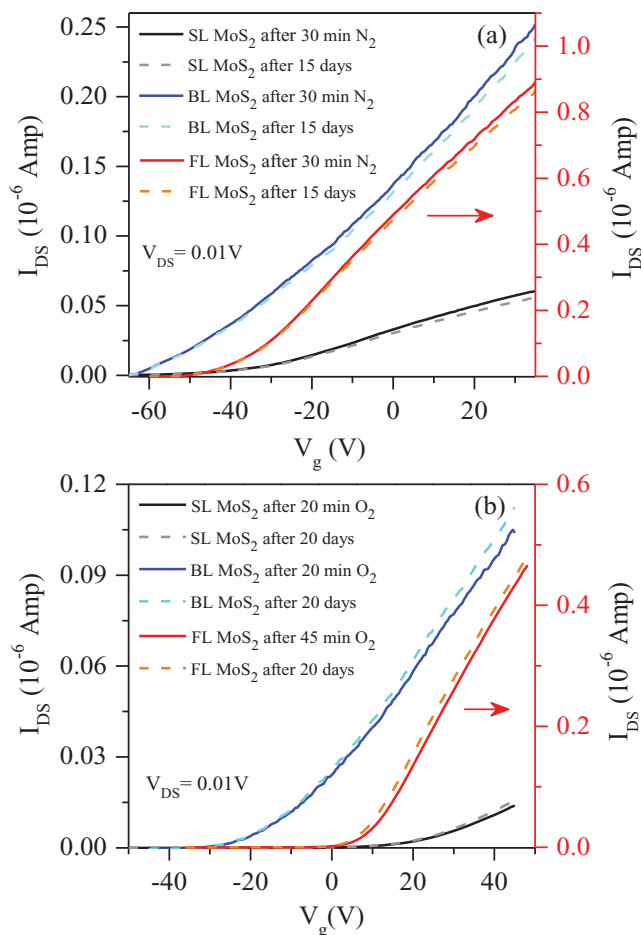


Figure 5. a) Stability of N_2 doping of SL, BL, and FL MoS_2 transistors as revealed by a I_{DS} - V_g plot after 30 min exposure to N_2 gas in the presence of DUV light and a plot for the same devices after 15 days. b) Stability of devices as revealed by a I_{DS} - V_g plot for SL, BL, and FL MoS_2 transistors after recovery with O_2 gas in the presence of DUV light and a plot for the same devices after 20 days.

each exposure to DUV and N_2 or O_2 gas, devices were characterized by Raman spectroscopy and measurements of electrical transport. The sample chamber was vacuum pumped for 30 min before each electrical measurement and during each measurement. All electrical measurements were thus made in a vacuum at room temperature.

Supporting Information

Supporting Information is available from the Wiley Online Library or from the author.

Acknowledgements

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